Additions and Corrections

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Benbrahim Achour, Judite Costa, Rita Delgado, Emmanuelle Garrigues, Carlos F. G. C. Geraldes, Nikolaus Korber, Françoise Nepveu, and M. Isabel Prata: Triethylenetetramine-*N*,*N*′,*N*″,*N*‴,*N*‴-hexaacetic Acid (TTHA) and TTHA-Bis(butanamide) as Chelating Agents Relevant to Radiopharmaceutical Applications.

Page 2732. Reference 14 should read as follows: Delgado, R.; Figueira, M. C.; Quintino, S. *Talanta* **1997**, *45*, 451.

Page 2734. Correct columns with TTHA/Ga(III) and TTHA/Fe(III) data appear below. Note that footnote *g* corresponds only to ref 14 and that footnote *h* is now included.

Table 4 (Corrected Columns Only). Protonation (log K_i^H) Constants of the Ligand TTHA and Its Stability Constants (log $K_{M_mL_lH_h}$) with Al^{3+}, Ga^{3+}, Fe^{3+}, and In^{3+} (T = 25.0 °C; I = 0.10 M $(CH_3)_4NNO_3$)

equil quotient	TTHA
[HL]/[L][H]	10.63(5);[a] 10.62[b]
$[H_2L]/[HL][H]$	9.46(2);[a] 9.49[b]
$[H_3L]/[H_2L][H]$	6.11(3);[a] 6.10[b]
$[H_4L]/[H_3L][H]$	4.04(4);[a] 4.06[b]
$[H_5L]/[H_4L][H]$	2.75(4);[a] 2.75[b]
$[H_6L]/[H_5L][H]$	2.34(7);[a] 2.3[b]
$[H_7L]/[H_6L][H]$	-*a*; 1.80[b]
$[H_3L]/[L][H]^3$	26.20;[a] 26.21[b]
$[H_4L]/[L][H]^4$	30.24;[a] 30.27[b]
[AlL]/[Al][L]	20.23(2);[a] 21.0;[b] 18.74;[d] 19.7[e]
[AlHL]/[AlL][H]	5.97(1);[a] 5.85[b,e]
[AlL]/[AlLOH][H]	-
$[Al_2L]/[AlL][Al]$	9.55(3);[a] 9.20;[b,e] 8.9[e]
$[Al_2L]/[Al_2LOH][H]$	4.68(5);[a] -
$[Al_2L]/[Al_2L(OH)_2][H]^2$	9.87(6);[a] 11.7[b,e]
[GaL]/[Ga][L]	27.75(7);[a] 28.21;[b] 23.60;[d] 15.1[f]
[GaHL]/[GaL][H]	4.8(1);[a] 5.30;[b] 4.52[f]
$[GaH_2L]/[GaHL][H]$	3.9(1);[a] 3.96;[b] 3.54 [f]
$[GaH_3L]/[GaH_2L][H]$	-[a]; 2.57[b]
[GaL]/[GaLOH][H]	9.43(4);[a] 9.64[b]
$[Ga_2L]/[GaL][Ga]$	12.40(9);[a] 10.0 [f]
$[Ga_2L]/[Ga_2LOH][H]$	3.25(9);[a] -
$[Ga_2L]/[Ga_2L(OH)_2][H]^2$	7.46(7);[a] -
[FeL]/[Fe][L]	27.66(4);[g] 26.8;[b,e] 29.4[h]
[FeHL]/[FeL][H]	7.49(2);[g] 7.55;[b] 7.60;[e] 7.51[h]
$[FeH_2L]/[FeHL][H]$	2.05(2);[g] 2.68;[b] 2.75;[e] 2.60[h]
[FeL]/[FeLOH][H]	-; 9.6[b,h]
$[Fe_2L]/[FeL][Fe]$	12.13(2);[g] 13.7[b,e]
$[Fe_2L]/[Fe_2LOH][H]$	2.11(3);[g] -
$[Fe_2L]/[Fe_2L(OH)_2][H]^2$	5.91(5);[g] 6.4;[b] 7.0[e]
[InL]/[In][L]	26.88(6);[g] 26.6[b]
[InHL]/[InL][H]	7.30(3)[g]
$[InH_2L]/[InHL][H]$	2.33(4)[g]
[InL]/[InLOH][H]	-
$[In_2L]/[InL][In]$	9.0(1)[g]
$[In_2L]/[In_2LOH][H]$	4.2(1)[g]
$[In_2L]/[In_2L(OH)_2][H]^2$	-

[a] This work. [b] T = 25.0 °C; I = 0.10 M, ref 23a. [c] Reference 7. [d] Reference 28. [e] Reference 29a. [f] Reference 29b. [g] Reference 14. [h] Reference 29c.

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